| QI | P Code: 105008 Reg.No.: | |
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| First B.Pharm (Ayurveda) Degree Examinations, October 2017 | | |
| PHARMACEUTICAL ANALYSIS | | |
| Tir | me: 3 Hours • Answer all questions • Answer all questions | ks: 100 |
| Es | | (2x10=20) |
| 1. | Explain about type of p ^m indicators used in compexometric titration. Disc mechanism and structure | cuss its |
| 2. | Explain in detail about neutralization curves in acid base titrations. | |
| Short Notes (10x5=50) | | |
| 3. | Universal indicators | |
| 4. | How to minimize errors | |
| 5. | Pharmaceutical buffers | |
| 6. | Explain the methods for expressing concentration | |
| 7. | Explain the theories of acid base indicators | |
| 8. | Polyprotic system | |
| 9. | Differentiate between Mohr's method and Volhardz method | |
| 10 |). How adsorption indicator works | |
| 11 | I. Explain about the principle behind the assay of boric acid | |
| 12 | 2. How halogen acid salt of weak base analyzed by non aqueous titration | |
| An | nswer Briefly | (10x3=30) |
| 13 | 3. Ideal properties for a primary standard | |
| 14. What is buffer capacity | | |
| 15. How phenolphthalein works | | |
| 16. Types of filter papers used in gravimetry | | |
| 17. Common ion effect | | |
| 18. Define oxidation and reduction | | |
| 19. Explain about conjugate acid base pairs | | |
| 20. Explain the construction of hydrogen electrode | | |

21. Preparation of 0.05m EDTA solution

22. Solvents used in non-aqueous titration.